

Computer prediction and synthesis of new azoles based on N-benzoyl-N'-(9,10-dioxo-9,10-dihydroanthracen-1-yl)thioureas

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New thiazole and 1,2,4-triazole derivatives of 9,10-anthraquinone were obtained by the reaction of N-benzoyl-N'-(9,10-dioxo-9,10-dihydroanthracene-1-yl)-thioureas with bromoacetone in the presence of triethylamine and hydrazine hydrate in chloroform, respectively. The PASS computer program was used to predict the biological activity spectra and to determine the most promising biological activities for experimental testing. Thus, it has been shown that the synthesized compounds are a promising class for the creation of substances with a wide range of biological activity.

Keywords: 9,10-anthraquinones, benzoylthioureas, cyclization, thiazol-2-imines, 1,2,4-triazoles

Introduction

The possibility of obtaining a large number of in practical terms valuable substances based on 9,10-anthraquinone derivatives is in accordance with the most important area of synthetic organic chemistry – the targeted synthesis of new organic compounds possessing a specified set of chemical, physical, and biological properties. An increasing number of publications about the synthesis and application of 9,10-anthraquinone derivatives testifies a continued interest in this series of compounds worldwide. Despite the widely studied chemistry of 9,10-anthraquinone, many classes of 9,10-anthraquinone derivatives remain scarcely explored.

On the other hand, N-arylthioureas have a strong synthetic potential. They are important reagents for the synthesis of several heterocycles, such as imidazolidinyl-2-thiones [1, 2], 2-aryliminothiazolines [3–5], 1,2,4-triazoles [6], 1,3-thiazines [7] and indeno[1,2-d][1,3]thiazepines [8]. 2-Iminothiazolines are characterized by a wide range of biological properties [9–12]. The thiazolidene-2-imine [13] or thiazol-2-imine [14] or 2-iminothiazoline [15–18] ring system, as it has been named by different groups, is present in several drug candidates possessing interesting biological activities such as muscarinomimetic, antimycotic, hypolipemic, antidiabetic,

thrombopoietinagonism, cell adhesion antagonists, platelet GPIIb/IIIa receptor antagonists, anti-inflammatory, analgesic and kinase (CDK1, CDK5 and GSK3) inhibition, schistosomicides, cardiotonics and trichomonides [19–23]. Thiazoline derivatives have been found to exhibit promising applications in agriculture, such as acaricides, insecticides, and plant growth regulators [24, 25]. Recently, 2-iminothiazolines have been found to show antifungal activity [26] and skin whitening action [27].

The 1*H*-1,2,4-triazole compounds show important pharmacological activities, e. g., antifungal and antiviral. Examples of such compounds bearing the 1,2,4-triazole residues are fluconazole [28] and ribavirin, the powerful azole antifungal agent as well as the potent antiviral *N*-nucleoside [29]. Furthermore, various 1,2,4-triazole derivatives have been reported as fungicidal [30], insecticidal [31], and antimicrobial [32] agents. Some of them have shown antitumor activity [33, 34]; anticonvulsants [35], antidepressants [36], plant growth regulators, and anticoagulants [37] have been identified among them as well.

The reaction of *N,N'*-disubstituted thioureas with α -bromoketones [5, 13] allows obtaining a variety of *N*-substituted 2-iminothiazoles. However, their derivatives that contain 9,10-dioxo-9,10-dihydroanthracenyl fragments at the 3rd position of the heterocycle remain unknown up to date. 1,2,4-Triazole derivatives of 9,10-anthraquinone are

also not described in the literature. Taking into account the known biological properties of anthraquinone derivatives [38–40], it seems appropriate to design hybrid structures containing anthraquinone and thiazole or triazole cycles.

Materials and methods

¹H NMR and ¹³C spectra were recorded with a Varian Mercury-400 spectrometer (400 and 100 MHz, respectively) at 25 °C in DMSO-d₆, using TMS as an internal standard. IR spectra were obtained on a Specord M80 spectrophotometer, using KBr tablets. Chromato-mass spectra were recorded on Agilent 110\DAD\HSD\VLG 119562 device. Melting points were determined with an automatic APA1 melting point apparatus and are uncorrected.

N-[3-(R¹,R²,R³)-9,10-Dioxo-9,10-dihydroanthracen-1-yl]-4-methylthiazol-2(3H)-ylidene]benzamide **2 a–e (general method). To a stirred suspension of 0.749 mmol of N-benzoylthiourea **1 a–e** and of 0.104 ml (0.749 mmol) of triethylamine in 30 ml of acetone, at stirring, the solution of 0.0384 ml (0.749 mmol) of bromine in 10 ml of acetone was added over 10 min. The reaction mixture was kept at room temperature for 2 h, the precipitate formed was filtered, washed with acetone and water, dried, and recrystallized from toluene.**

N-(3-(9,10-Dioxo-9,10-dihydroanthracen-1-yl)-4-methylthiazol-2(3H)-ylidene)benzamide (2 a).

Yield 68 %, Mp. 210–211 °C.

¹H NMR, δ, ppm (*J*, Hz): 1.99 (3H, s, CH₃); 6.92 (1H, s, CH=); 6.98 (3H, m, H Ar); 7.68–7.70 (3H, m, H Ph); 7.99 (2H, m, H Ph); 8.21 (3H, m, H Ar); 8.53 (1H, d, *J* = 8.4, H Ar).

¹³C NMR, δ, ppm: 14.2 (CH₃CH=); 106.7 (CH=); 126.8, 127.3, 127.5, 128.4 (C Ar); 128.7, 129.1, 129.3, 129.1 (C Ph); 131.5 (C Ar); 132.4 (C Ph); 133.5, 133.6, 134.4, 134.9, 135.2 (C Ar); 135.8 (C Ph); 137.8 (C-N); 143.3 (CH₃CH=); 168.5 (C=N); 173.5 (COPh); 181.8 (CO); 182.5 (CO).

IR spectra, ν, cm⁻¹: 1680, 1631 (C=O quinone ring), 1685 (COPh), 1471 (C=N).

Chromato-mass spectra, *m/z* (*I*_{rel}, %): 425 [M+H]⁺ (69).

Found, %: C 70.61; H 4.01; N 6.71; S 7.79. C₂₅H₁₆N₂O₃S.
Calculated, %: C 70.74; H 3.80; N 6.60; S 7.55.

N-(4-Methyl-3-(2-methyl-9,10-dioxo-9,10-dihydroanthracen-1-yl)thiazol-2(3H)-ylidene)benzamide (2 b).

Yield 51 %, Mp. 206 °C.

¹H NMR, δ, ppm (*J*, Hz): 1.99 (3H, s, CH₃); 2.17 (3H, s, CH₃); 7.01 (1H, s, CH=); 7.24–7.27 (2H, m, H Ph); 7.36 (1H, m, H Ph); 7.72 (2H, m, H Ph); 7.85–7.90 (2H, m, H Ar); 8.02 (1H, d, *J* = 7.5, H Ar); 8.14 (1H, d, *J* = 7.5, H Ar); 8.21 (1H, d, *J* = 7.5, H Ar); 8.44 (1H, d, *J* = 7.6, H Ar).

¹³C NMR, δ, ppm: 13.9 (CH₃CH=); 17.7 (CH₃); 106.1 (CH=); 126.8, 127.3, 127.6, 128.5 (C Ar); 128.8, 128.9, 129.0, 129.1 (C Ph); 131.8 (C Ar); 132.8 (C Ph); 133.7, 133.9, 134.0, 135.0, 135.1 (C Ar); 135.2 (C Ph); 137.1 (C Ar); 137.3 (C-N); 144.9 (CH₃CH=); 168.7 (C=N); 173.1 (COPh); 182.1 (CO); 182.4 (CO).

IR spectra, ν, cm⁻¹: 1681, 1623 (C=O quinone ring), 1670 (COPh), 1397 (C=N).

Chromato-mass spectra, *m/z* (*I*_{rel}, %): 439 [M+H]⁺ (97).

Found, %: C 71.61; H 4.01; N 6.71; S 7.79. C₂₆H₁₈N₂O₃S.
Calculated, %: C 71.22; H 4.14; N 6.39; S 7.51.

N-(3-(4-Amino-9,10-dioxo-9,10-dihydroanthracen-1-yl)-4-methylthiazol-2(3H)-ylidene)benzamide (2 c).

Yield 50 %, Mp. 296–297 °C. ¹H NMR, δ, ppm (*J*, Hz): 1.98 (3H, s, CH₃); 6.89 (1H, s, CH=); 7.26 (3H, m, H Ar, NH₂); 7.38–7.41 (2H, m, H Ar); 7.49–7.57 (3H, m, H Ph); 7.68–7.73 (2H, m, H Ph); 7.87–7.91 (2H, m, H Ar); 8.23 (1H, d, *J* = 7.6, H Ar).

¹³C NMR, δ, ppm: 13.7 (CH₃CH=); 107.2 (CH=); 114.5, 125.3, 126.1, 127.2 (C Ar); 128.1, 128.2, 129.3, 129.4 (C Ph); 130.1 (C-N); 130.9, 131.3 (C Ar); 132.3 (C Ph); 133.1, 133.2, 133.5, 133.8 (C Ar); 136.1 (C Ph); 139.8 (CH₃CH=); 147.8 (C-NH₂); 169.7 (C=N); 174.5 (COPh); 183.1 (CO); 184.3 (CO).

IR spectra, ν, cm⁻¹: 3365 (NH₂), 1683, 1625 (C=O quinone ring), 1650 (CONH), 1450 (C=N).

Chromato-mass spectra, *m/z* (*I*_{rel}, %): 440 [M+H]⁺ (73).

Found, %: C 68.61; H 4.11; N 9.69; S 7.57. C₂₅H₁₇N₃O₃S.
Calculated, %: C 68.32; H 3.90; N 9.56; S 7.29.

N-(3-(5-Amino-9,10-dioxo-9,10-dihydroanthracen-1-yl)-4-methylthiazol-2(3H)-ylidene)benzamide (2 d).

Yield 69 %, Mp. 232 °C.

¹H NMR, δ, ppm (*J*, Hz): 1.97 (3H, s, CH₃); 6.92 (1H, s, CH=); 7.15–7.26 (5H, m, H Ar); 7.34–7.46 (2H, m, H Ph); 7.68 (3H, m, H Ph, NH₂); 7.88 (1H, m, H Ar); 8.13–8.16 (1H, m, H Ar); 8.53 (1H, d, *J* = 8.4, H Ar).

¹³C NMR, δ, ppm: 13.8 (CH₃CH=); 106.8 (CH=); 111.9, 115.4, 120.1, 122.2, 123.7 (C Ar); 128.1, 129.2, 129.3, 129.4, 132.4 (C Ph); 132.9, 133.4, 135.0, 135.3, 135.7 (C Ar); 136.1 (C Ph); 138.2 (C-N); 139.3 (CH₃CH=); 151.8 (C-NH₂); 168.2 (C=N); 174.5 (COPh); 183.6 (CO); 184.4 (CO).

Chromato-mass spectra, *m/z* (*I*_{rel}, %): 440 [M+H]⁺ (82).

N-(3-(4-Benzamido-9,10-dioxo-9,10-dihydroanthracen-1-yl)-4-methylthiazol-2(3H)-ylidene)benzamide (2 e).

Yield 48 %, Mp. 198 °C.

¹H NMR, δ, ppm (*J*, Hz): 2.04 (3H, s, CH₃); 6.95 (1H, s, CH=); 7.36 (1H, m, H Ar); 7.72–8.27 (13H, m, H Ar); 8.88 (2H, m, H Ar); 9.36 (2H, m, H Ar); 13.30 (1H, s, NH).

¹³C NMR, δ, ppm: 14.1 (CH₃CH=); 106.3 (CH=); 125.1, 126.5 (C Ar); 128.0, 128.1 (C Ph); 118.2 (C Ar); 128.8, 128.9, 129.4, 129.3 (C Ph); 136.2 (C Ar); 127.3, 127.2, 132.4, 133.3 (C Ph); 130.3 (C Ar); 132.5 (C Ph); 134.3, 133.7 (C Ar); 138.2 (C-N); 133.4, 133.0 (C Ar); 136.0 (C Ph); 129.5 (C Ar); 133.2 (C-NH); 138.8 (CH₃CH=); 165.7 (CO); 167.8 (C=N); 174.4 (COPh); 182.6 (CO); 184.5 (CO).

IR spectra, ν, cm⁻¹: 3345 (NHCO), 1683, 1625, 1661 (C=O quinone ring), 1650 (CONH), 1450 (C=N).

Found, %: C 70.61; H 3.81; N 7.69; S 7.77. C₃₂H₂₁N₃O₄S.
Calculated, %: C 70.70; H 3.89; N 7.73; S 5.90.

R^1, R^2, R^3 -[(5-Phenyl-4H-1,2,4-triazol-3-yl)amino]-anthracene-9,10-diones **3 a–e** (general method). To a stirred suspension of 0.0518 mmol of N-benzoylthiourea **1 a–e** in 30 ml of chloroform, 0.013 g (0.259 mmol) of hydrazine hydrate was added. The reaction mixture was heated under reflux for 5 h and cooled to room temperature. The precipitate formed was filtered, washed with water, and dried.

1-[(5-Phenyl-4H-1,2,4-triazol-3-yl)amino]-anthracene-9,10-dione (3 a).

Yield 81 %, Mp. 248 °C.

$^1\text{H NMR}$, δ , ppm (J , Hz): 7.52–7.67 (5H, m, H Ar); 7.85–8.21 (7H, m, H Ar); 8.88 (1H, d, $J = 8.7$, H Ar); 12.15 (1H, s, NH); 14.16 (1H, s, NH).

IR spectra, ν , cm^{-1} : 3286 (NH triazole ring), 1683, 1622 (C=O quinone ring).

Chromato-mass spectra, m/z (I_{rel} , %): 367 $[\text{M}+\text{H}]^+$ (100).

Found, %: C 72.17; H 4.01; N 15.37. $\text{C}_{22}\text{H}_{14}\text{N}_4\text{O}_2$.

Calculated, %: C 72.12; H 3.85; N 15.29.

2-Methyl-1-[(5-phenyl-4H-1,2,4-triazol-3-yl)amino]anthracene-9,10-dione (3 b).

Yield 78 %, Mp. 226 °C.

$^1\text{H NMR}$, δ , ppm (J , Hz): 2.19 (3H, s, CH_3); 7.55–7.79 (2H, m, CH_{Ar}); 7.83–7.86 (2H, m, CH_{Ar}); 7.95–7.98 (3H, m, CH_{Ar}); 8.19–8.21 (2H, m, CH_{Ar}); 8.91 (1H, d, $J = 7.48$, H Ar); 12.69 (1H, s, NH); 14.61 (1H, s, NH).

IR spectra, ν , cm^{-1} : 3290 (NH triazole ring), 1685, 1625 (C=O quinone ring).

Found, %: C 72.69; H 4.14; N 14.68. $\text{C}_{23}\text{H}_{16}\text{N}_4\text{O}_2$.

Calculated, %: C 72.62; H 4.24; N 14.73.

1-Amino-4-[(5-phenyl-4H-1,2,4-triazol-3-yl)amino]anthracene-9,10-dione (3 c).

Yield 74 %, Mp. 273 °C.

$^1\text{H NMR}$, δ , ppm (J , Hz): 7.12–7.28 (4H, m, CH_{Ar} , NH_2); 7.58–7.92 (8H, m, CH_{Ar}); 8.40 (1H, d, $J = 7.48$ Hz, CH_{Ar}); 10.43 (1H, s, NH); 14.43 (1H, s, NH).

IR spectra, ν , cm^{-1} : 3361, 3315 (NH_2), 3304 (NH triazole ring), 1679, 1620 (C=O quinone ring).

Found, %: C 69.11; H 3.84; N 18.27. $\text{C}_{22}\text{H}_{15}\text{N}_5\text{O}_2$.

Calculated, %: C 69.28; H 3.96; N 18.36.

1-Amino-5-[(5-phenyl-4H-1,2,4-triazol-3-yl)amino]anthracene-9,10-dione (3 d).

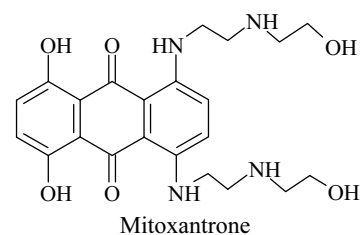
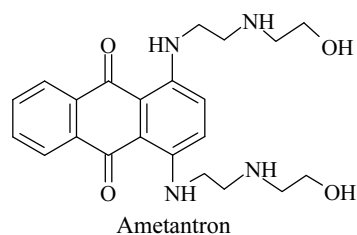
Yield 60 %, Mp. 260 °C.

$^1\text{H NMR}$, δ , ppm (J , Hz): 7.08 (2H, m, NH_2); 7.36–7.56 (5H, m, CH_{Ar}); 7.81–8.19 (5H, m, CH_{Ar}); 8.89 (1H, d, $J = 8.0$ Hz, CH_{Ar}); 11.07 (1H, s, NH); 13.36 (1H, s, NH).

IR spectra, ν , cm^{-1} : 3359, 3312 (NH_2), 3310 (NH triazole ring), 1684, 1625 (C=O quinone ring).

Found, %: C 69.19; H 3.90; N 18.29. $\text{C}_{22}\text{H}_{15}\text{N}_5\text{O}_2$.

Calculated, %: C 69.28; H 3.96; N 18.36.



N-(9,10-Dioxo-4-[(5-phenyl-4H-1,2,4-triazol-3-yl)amino]-9,10-dihydroanthracen-1-yl)benzamide (3 e).

Yield 70 %, Mp. 234 °C.

$^1\text{H NMR}$, δ , ppm (J , Hz): 7.39–7.69 (5H, m, CH_{Ar}); 7.87–7.97 (9H, m, CH_{Ar}); 8.71 (1H, d, $J = 8.9$ Hz, CH_{Ar}); 12.37 (1H, s, NH); 13.03 (1H, s, NH); 14.03 (1H, s, NH).

IR spectra, ν , cm^{-1} : 3328 (NH triazole ring), 1681, 1623 (C=O quinone ring), 1648 (CONH).

Found, %: C 71.61; H 4.05; N 14.31. $\text{C}_{29}\text{H}_{19}\text{N}_5\text{O}_3$.

Calculated, %: C 71.74; H 3.94; N 14.43.

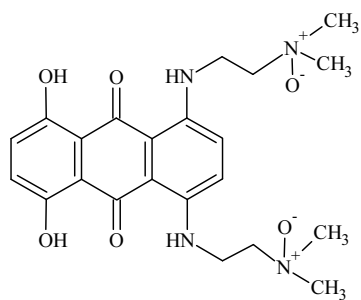
Results and discussion

Computer prediction of biological activity spectra using the PASS software

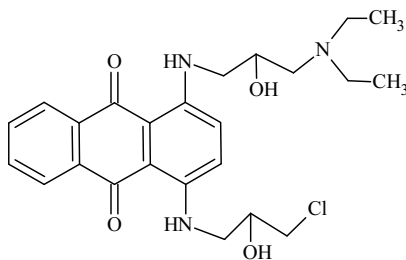
The prediction of the biological activity spectra of the synthesized compounds was performed employing the PASS computer program (Prediction of Activity Spectra for Substances) [41, 42]. The latest PASS version (2012.11.22) predicts 6400 kinds of biological activity based on the analysis of the training set including information about ~330000 drugs, drug-candidates, lead compounds, etc. The average prediction accuracy estimated in a leave-one-out cross-validation procedure for the whole training set is about 95 %. The PASS online version is freely available for a scientific community on the website [43]. Based on the PASS predictions, new pharmaceutical agents from diverse chemical classes with various kinds of biological activity have been discovered [43–45]. The PASS output is presented by a list of probable activities with two estimated probabilities: P_a – the probability to be “active”, and P_i – the probability to be “inactive”.

The higher the P_a value, the lower is the predicted probability of obtaining false positives in biological testing. For example, if one selects for testing only compounds for which a particular activity is predicted with $P_a > 0.9$, the expected probability to find inactive compounds in the selected set is very low; however, about 90 % of active compounds are missed. If one lowers the P_a threshold to 0.8, the probability to find inactive compounds is still low, but (only) about 80 % of active compounds are missed, etc. PASS uses the criteria $P_a = P_i$ as the default threshold, i. e. all compounds with $P_a > P_i$ are declared as being active.

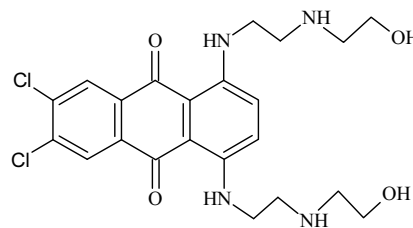
The computer prediction using the PASS Online program for such well-known anticancer drugs based on aminoderivatives of 9,10-anthraquinone as Ametantron, Mitoxantrone, Banoxantrone, and compounds on preclinical studies NSC-639365 and M-18 [46] was carried out. Results are in a complete agreement with the experimental data on anticancer activity (Table 1).



Banoxantron



NSC-639365



M-18

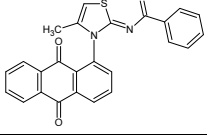
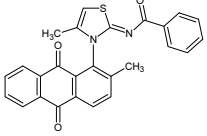
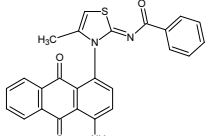
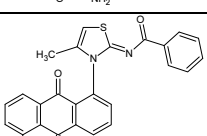
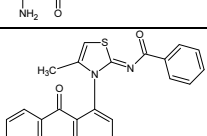
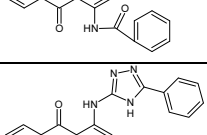
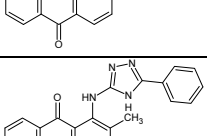
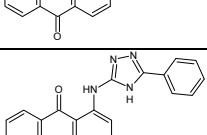
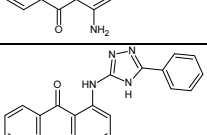
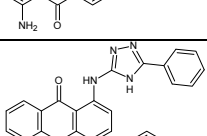
An example of the predicted pharmacological effects for Ametantron is given below (only data with $P_a > 0.5$ are shown).

P_a	P_i	Pharmacological effects
0.742	0.004	Radiosensitizer
0.745	0.019	Antineoplastic
0.728	0.003	Peroxidase substrate
0.715	0.006	DNA-(apurinic or apyrimidinic site) lyase inhibitor
0.701	0.005	Choline-phosphate cytidylyltransferase inhibitor
0.697	0.011	3-Hydroxybenzoate 6-monooxygenase inhibitor
0.662	0.006	Antineoplastic (non-Hodgkin's lymphoma)
0.708	0.072	Phobic disorders treatment
0.635	0.005	Alkylator
0.640	0.019	Manganese peroxidase inhibitor
0.663	0.064	CDP-glycerol glycerophosphotransferase inhibitor
0.607	0.011	CYP2E substrate
0.586	0.002	Beta-Lysine 5,6-aminomutase inhibitor
0.584	0.008	Chemosensitizer
0.657	0.083	Gluconate 2-dehydrogenase (acceptor) inhibitor
0.643	0.090	Ubiquinol-cytochrome-c reductase inhibitor
0.569	0.023	Fibrolase inhibitor
0.560	0.017	Cytostatic
0.602	0.062	CYP2J2 substrate
0.578	0.039	Glucose oxidase inhibitor
0.613	0.082	CYP2J substrate
0.532	0.003	Acetylenecarboxylate hydratase inhibitor
0.597	0.074	CYP2C12 substrate
0.560	0.036	General pump inhibitor
0.527	0.008	Antineoplastic (multiple myeloma)
0.521	0.005	DNA intercalator
0.522	0.006	RNA synthesis inhibitor
0.546	0.031	Hydrogen dehydrogenase inhibitor
0.545	0.030	Superoxide dismutase inhibitor
0.555	0.041	Aldehyde oxidase inhibitor
0.556	0.044	Macrophage colony stimulating factor agonist
0.500	0.003	Antineoplastic, alkylator

Table 1. Predicted antineoplastic (anticancer) activity using the PASS program for known derivatives of 9,10-anthraquinone with anticancer activity

Compound	Activity P_a	Antineoplastic	Antineoplastic (non-Hodgkin's lymphoma)	Antineoplastic (multiple myeloma)
Ametantron		0.745	0.662	0.527
Mitoxantrone		0.798	0.671	0.536
Banoxantron		0.906	0.868	0.366
NSC-639365		0.737	0.547	0.859
M-18		0.646	0.641	0.521

Table 2. Typical activities predicted for new 2-iminothiazole **2 a–e** and 1,2,4-triazole **3 a–e** derivatives of 9,10-anthraquinone ($P_a > 0.4$)

Compound Activity (P_a)	Mucomembranous protector	Transcription factor STAT3 inhibitor	Antineoplastic (non-Hodgkin's lymphoma)	Muramoyltetrapeptide carboxypeptidase inhibitor	Antineoplastic	Angiogenesis inhibitor	3-Hydroxybenzoate 6-monoxygenase inhibitor	Protein kinase inhibitor	Pterin deaminase inhibitor	Antiarthritic
 2a	0.684	0.508	-	0.596	-	-	-	-	-	-
 2b	0.648	0.500	-	0.566	-	-	-	-	-	-
 2c	0.421	-	0.430	0.527	-	-	-	-	-	-
 2d	0.421	-	0.430	0.527	-	-	-	-	-	-
 2e	0.451	0.499	-	0.486	-	-	-	-	-	-
 3a	0.684	-	0.436	-	0.810	0.885	0.834	0.443	0.589	0.500
 3b	0.648	-	0.411	-	0.774	0.813	0.684	0.557	0.545	0.441
 3c	0.421	-	0.491	-	0.811	0.846	0.708	0.434	0.553	0.432
 3d	-	-	0.491	-	0.811	0.846	0.708	0.434	0.553	0.432
 3e	-	-	0.456	-	0.781	0.864	0.646	0.401	-	0.558

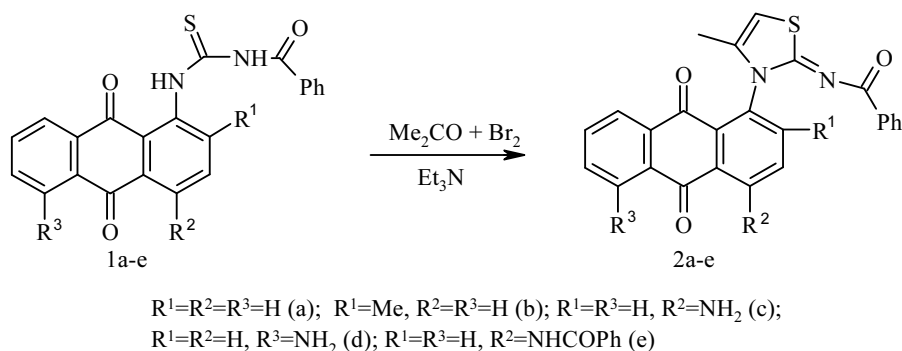
Note: “-” predictive activity is absent when $P_a > 0.4$.

The data obtained in experimental biological testing and calculated using the computer on-line PASS program provided the basis for predicting probable spectrum of anticancer activity of a series of new azole derivatives of 9,10-anthraquinone (Table 2).

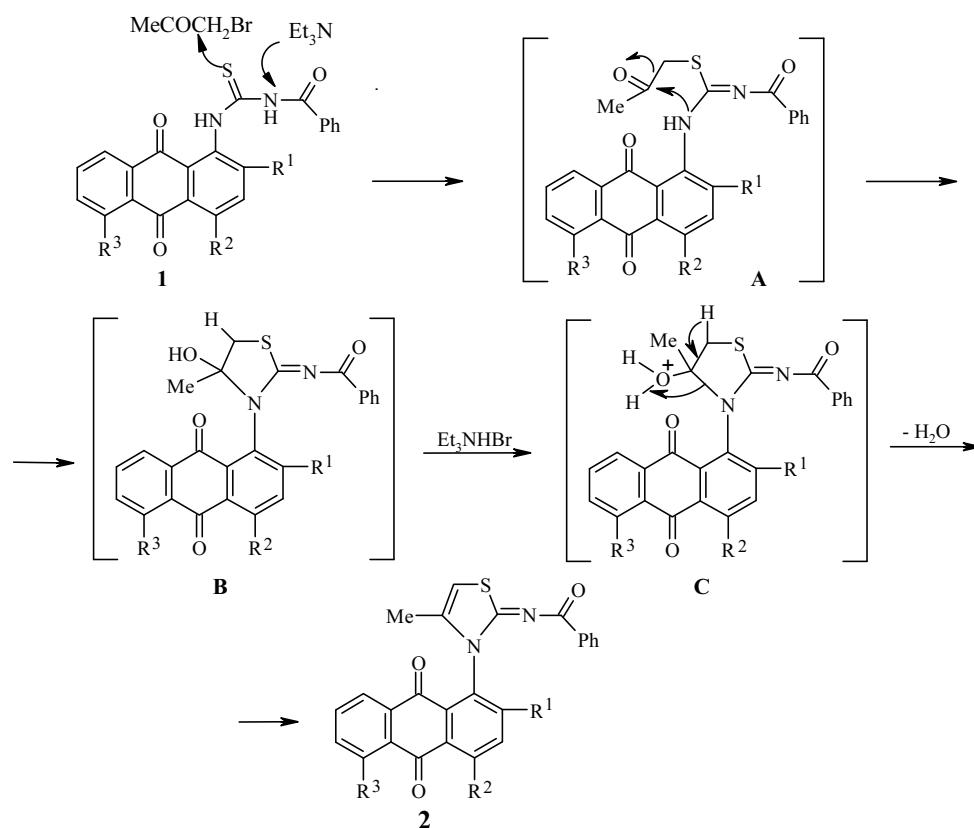
Some predicted typical pharmacological effects for ten new compounds – 2-iminothiazoles **2 a-e** and 1,2,4-triazoles **3 a-e** – are shown in Table 2 ($Pa > 0.4$), which include: *mucomembranous protector, transcription factor STAT3 inhibitor, antineoplastic, muramoyl-tetrapeptide carboxypeptidase inhibitor, angiogenesis inhibitor, 3-hydroxybenzoate 6-monoxygenase inhibitor, protein kinase inhibitor, pterin deaminase inhibitor, antiarthritic*, etc. In the view of these data calculated using the PASS, the synthesis of new biheterocyclic structures, including 9,10-anthraquinone, and thiazole or triazole fragments seems reasonable.

Chemistry

Benzoylthioureas **1 a-e** synthesized as described

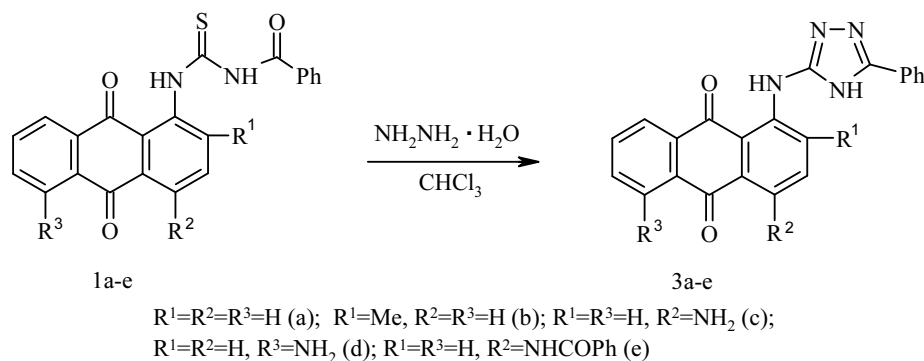


Scheme 1. Synthesis of N-[3-(R^1,R^2,R^3)-9,10-dioxo-9,10-dihydroanthracen-1-yl]-4-methylthiazol-2(3H)-ylidene]benzamides **2 a-e**



Scheme 2. Mechanism of the formation of N-[3-(R^1,R^2,R^3)-9,10-dioxo-9,10-dihydroanthracen-1-yl]-4-methylthiazol-2(3H)-ylidene]benzamides **2 a-e**

In the ^1H NMR spectra of compounds **2 a–e**, along with the signals of aromatic protons, singlets of H-5 protons of the thiazole cycle (6.89–7.01 ppm) and methyl groups (1.97–2.04 ppm) are present. The formation of thiazole cycle was reliably confirmed by the ^{13}C NMR spectra displaying the characteristic singlets of C^4 , C^5 and C^2 atoms at 106.1–107.2, 139.3–144.9, and 168.2–169.7 ppm, respectively.



Scheme 3. Synthesis of $\text{R}^1, \text{R}^2, \text{R}^3$ -[(5-phenyl-4*H*-1,2,4-triazol-3-yl)amino]anthracen-9,10-diones **3 a–e**

In the IR spectra of the synthesized compounds, characteristic absorption bands of two carbonyl groups of an anthraquinone fragment within the $1620\text{--}1685\text{ cm}^{-1}$ and the ones for the NH group of the triazole cycle at $3304\text{--}3328\text{ cm}^{-1}$ are present. In the ^1H NMR spectra of the triazole derivatives **3 a–e**, the singlets of the aminotriazole cycle proton are present at 14.03–14.63 ppm.

Conclusions

New potentially biologically active *N*-(9,10-dioxo-9,10-dihydroanthracene-1-yl)-2-(*N*-benzoylimino)thiazoles and [(5-phenyl-4*H*-1,2,4-triazol-3-yl)amino]anthracene-9,10-diones were obtained by the reaction of *N*-benzoyl-*N'*-(9,10-dioxo-9,10-dihydroanthracene-1-yl)thioureas with bromoacetone and hydrazine hydrate, respectively. The mechanism of *N*-[3-($\text{R}^1, \text{R}^2, \text{R}^3$)-9,10-dioxo-9,10-dihydroanthracen-1-yl]-4-methylthiazol-2(3*H*)-ylidene]benzamides formation was proposed. The biological activity spectra of the synthesized compounds have been estimated by using the PASS computer program.

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References

- Hartung J., Rosenbaum K., Beyer L., Fernandes V. Zur reversiblen umwandlung von 3-diethylamino-5-phenyl-1,2,4-dithiazolium-halometallaten in metallchelat der 3-(thio)benzoyl(thio)harnstoffe // J. Prakt. Chem. 1991. Vol. 333. P. 537–544.
- Zeng R.-S., Zou J.-P., Zhi S.-J., Chen J., Shen Q. Aldol- and Mannich-type reactions via *in situ* olefin migration in ionic liquid // Org. Lett. 2003. Vol. 5. P. 657–660.
<http://dx.doi.org/10.1021/ol0273102>
- Manaka A., Ishii T., Takahashi K. 2-Acylimino-3-alkyl-3*H*-thiazoline derivatives: one-pot, three-component condensation synthesis of novel β -turn mimics // Tetrahedron Lett. 2005. Vol. 46. N 3. P. 419–422.
<http://dx.doi.org/10.1016/j.tetlet.2004.11.105>
- Wang X., Wang F., Quan Z., Zhang Z., Wang M. One-pot synthesis of 2-acylimino-3-arylthiazoline derivatives in aqueous media // Synt. Commun. 2006. Vol. 36. P. 2453–2460.
<http://dx.doi.org/10.1080/00397910600781208>
- Murru S., Singh C. B., Kalava V., Patel B. K. A convenient one-pot synthesis of thiazol-2-imines: application in the construction of pifithrin analogues // Tetrahedron. 2008. Vol. 64. P. 1931–1942.
<http://dx.doi.org/10.1016/j.tet.2007.11.076>
- Kodomari M., Suzuki M., Tanigawa K., Aoyoma T. A convenient and efficient method for the synthesis of mono- and *N,N*-disubstituted thioureas // Tetrahedron Lett. 2005. Vol. 46. P. 5841–5843.
<http://dx.doi.org/10.1016/j.tetlet.2005.06.135>
- Aly A. A., Ahmed E. K., El-Mokadam K. M. Reactions of arylthioureas with acetylenic esters and dibenzoyl ethylene. Selectivity towards the formation of new 1,3-thiazines // J. Heterocycl. Chem. 2007. Vol. 44. P. 1431–1438.
<http://dx.doi.org/10.1002/jhet.5570440630>
- Aly A. A., Brown A. B., Ramadan M., Abdel-Aziz M., Abuo-Rahma G., Radwan M. F., Gamal-Eldeen A. M. Selectivity of *N*-aroyl-*N'*-arylthioureas towards 2-(1,3-dioxo-1*H*-inden-2(3*H*)-ylidene)malononitrile. New synthesis of (*Z*)-*N*-((*E*)-4-amino-1-aryl-5-cyano-6-oxo-1*H*-indeno[1,2-*d*][1,3]-thiazepin-2(6*H*)-ylidene)-4-arylamides of antitumor and antioxidant activities // J. Heterocycl. Chem. 2010. Vol. 47. N 3. P. 503–508.
<http://dx.doi.org/10.1002/jhet.344>

The synthesis of novel 1,2,4-triazole derivatives of 9,10-anthraquinone was carried out by the procedure described in [6]. Benzoylthioureas **1 a–e** were heated with hydrazine hydrate in a ratio of 1 : 5 in refluxing chloroform. As a result, triazolylaminoanthracenes **3 a–e** were obtained with 60–81 % yields (Scheme 3).

9. **Huang D, Ding Y, Luo W. M, Bender S, Qian C. N, Kort E, Zhang Z. F, VandenBeldt K, Duesbery N. S, Resau J. H, Teh B. T.** Inhibition of MAPK kinase signaling pathways suppressed renal cell carcinoma growth and angiogenesis *in vivo* // *Cancer Res.* 2008. Vol. 68. P. 81–88.
<http://dx.doi.org/10.1158/0008-5472.CAN-07-5311>.
10. **Tomizawa M., Kagabu S., Ohno I., Durkin K. A., Casida J. E.** Potency and selectivity of trifluoroacetylrimino and pyrazinoylimino nicotinic insecticides and their fit at a unique binding site niche // *J. Med. Chem.* 2008. Vol. 51. P. 4213–4218.
<http://dx.doi.org/10.1021/jm800191a>
11. **Zhang N., Tomizawa M., Casida J. E.** Structural features of azidopyridinyl neonicotinoid probes conferring high affinity and selectivity for mammalian $\alpha 4\beta 2$ and drosophila nicotinic receptors // *J. Med. Chem.* 2002. Vol. 45. P. 2832–2840.
<http://dx.doi.org/10.1021/jm010508s>
12. **Walczynski K., Guryn R., Zuiderveld O. P., Zhang M.-Q., Timmerman H.** Histamine H1 receptor ligands: Part II. Synthesis and *in vitro* pharmacology of 2-[2-(phenylamino)thiazol-4-yl]ethanamine and 2-(2-benzhydrylthiazol-4-yl)ethanamine derivatives // *Farmaco.* 2000. Vol. 55. P. 569–574.
[http://dx.doi.org/10.1016/S0014-827X\(00\)00087-2](http://dx.doi.org/10.1016/S0014-827X(00)00087-2)
13. **Singh C. B., Murru S., Kavala V., Patel B. K.** It is “thiazolidene-2-imine” and not imidazole-2-thione as the reaction product of 1-benzoyl-3-phenylthiourea with Br_2 /enolizable ketone // *Org. Lett.* 2006. Vol. 8. N 23. P. 5397–5399.
<http://dx.doi.org/10.1021/ol062371b>
14. **Singh C. B., Murru S., Kavala V., Patel B. K.** 3-Aryl-1-benzoylthioureas with α -bromoketones in water form 2-N-benzoyl-3-arylthiazol-2(3H)-imines, not 3-aryl-1-benzoylimidazole-2-thiones // *J. Chem. Res.* 2007. Vol. 3. P. 136–137.
<http://dx.doi.org/10.3184/030823407X196719>
15. **De Kimpe N., Boelens M., Declereq J.-P.** A Novel synthesis of 2-imino-4-thiazolines via α -bromoketimines // *Tetrahedron.* 1993. Vol. 49. N 16. P. 3411–3424.
[http://dx.doi.org/10.1016/S0040-4020\(01\)90168-1](http://dx.doi.org/10.1016/S0040-4020(01)90168-1).
16. **D’hooghe M., Waterinckx A., De Kimpe N.** A novel entry toward 2-imino-1,3-thiazolidines and 2-imino-1,3-thiazolines by ring transformation of 2-(thiocyanomethyl)aziridines // *J. Org. Chem.* 2005. Vol. 70. P. 227–232.
<http://dx.doi.org/10.1021/jo048486f>
17. **Kasmi S., Hamelin J., Benhaoua H.** Microwave-assisted solvent-free synthesis of iminothiazolines // *Tetrahedron Lett.* 1998. Vol. 39. P. 8093–8096.
[http://dx.doi.org/10.1016/S0040-4039\(98\)01836-X](http://dx.doi.org/10.1016/S0040-4039(98)01836-X)
18. **Zhu X. X., Yu Q. S., Culter R. G., Culmsee C. W., Holloway H. W., Lahiri D. K., Mattson M. P., Greig N. H.** Novel p53 inactivators with neuroprotective action: Syntheses and pharmacological evaluation of 2-imino-2,3,4,5,6,7-hexahydrobenzothiazole and 2-imino-2,3,4,5,6,7-hexahydrobenzoxazole derivatives // *J. Med. Chem.* 2002. Vol. 45. P. 5090–5097.
<http://dx.doi.org/10.1021/jm020044d>
19. **Ivanov Y. Y., Tkachenko S. E., Proshin A. N., Bachurin S. O.** Derivatives of 2-amino-2-thiazoline – novel chemotype with muscarinomimetic activity // *Biomed. Khim.* 2003. Vol. 49. N 1. P. 92–95.
20. **Baumann R. J., Mayer G. D., Fite L. D., Gill L. M., Harrison B. L.** *In vitro* and *in vivo* candidal activities of 2-(p-n-hexylphenylamino)-1,3-thiazoline // *Chemotherapy* 1991. Vol. 37. P. 157–165.
<http://dx.doi.org/10.1159/000238849>
21. **Sondhi S. M., Singh N., Lahoti A. M., Bajaj K., Kumar A., Lozach O., Meijer L.** Synthesis of acridinyl-thiazolino derivatives and their evaluation for anti-inflammatory, analgesic and kinase inhibition activities // *Bioorg. Med. Chem.* 2005. Vol. 13. P. 4291–4299.
<http://dx.doi.org/10.1016/j.bmc.2005.04.017>
22. **Matsuda H., Shimoda H., Yoshikawa M.** Dimeric sesquiterpene thioalkaloids with potent immunosuppressive activity from the rhizome of *Nuphar pumilum*: structural requirements of *Nuphar* alkaloids for immunosuppressive activity // *Bioorg. & Med. Chem.* 2001. Vol. 9. N 4. P. 1031–1035.
[http://dx.doi.org/10.1016/S0968-0896\(00\)00327-8](http://dx.doi.org/10.1016/S0968-0896(00)00327-8)
23. **Webel L. M., Degnam M. B., Harger G. F., Capps D. B., Islip P. J., Closier M. D.** Terpene compounds as drugs. 11. Anabolic 19-nortestosterone terpenoates // *J. Med. Chem.* 1972. Vol. 15. P. 995–998.
<http://dx.doi.org/10.1021/jm00279a031>
24. **Nagasaki F., Yamada T., Takahashi E., Kitagawa Y., Hatano R.** *Jpn. Kokai Tokkyo Koho JP 63250371*, 1988.
25. **Hoelzel H., Creuzburg D., Stohr P., Dehne H., Teller J., Kranz L., Luthardt H., Roethling T., Kaestner A.** *Ger. DD 258168*, 1988.
26. **Bae S., Hahn H. G., Nam K. D., Mah H.** Solid-phase synthesis of fungitoxic 2-imino-1,3-thiazolines // *J. Comb. Chem.* 2005. Vol. 7. P. 7–9.
<http://dx.doi.org/10.1021/cc049854w>
27. **Kim D.-S., Joeng Y.-M., Park I.-K., Hahn H.-G., Lee H.-K., Kwon S.-B., Jeong J. H., Yang S. J. Sohn U. D., Park K.-C.** A new 2-imino-1,3-thiazoline derivative, KHG22394, inhibits melanin synthesis in mouse B16 melanoma cells // *Biol. Pharm. Bull.* 2007. Vol. 30. P. 180–183.
<http://dx.doi.org/10.1248/bpb.30.180>
28. **Tsukuda T., Shiratori Y., Watanabe M., Otsuka H., Hattori K., Shirai M., Shimma N.** Modeling, synthesis and biological activity of novel antifungal agents (1) // *Bioorg. Med. Chem. Lett.* 1998. Vol. 8. P. 1819–1824.
[http://dx.doi.org/10.1016/S0960-894X\(98\)00316-3](http://dx.doi.org/10.1016/S0960-894X(98)00316-3)
29. **Witkoaski J. T., Robins R. K., Sidwell R. W., Simon L. N.** Design, synthesis, and broad spectrum antiviral activity of 1- β -D-ribofuranosyl-1,2,4-triazole-3-carboxamide and related nucleosides // *J. Med. Chem.* 1972. Vol. 15. P. 1150–1154.
<http://dx.doi.org/10.1021/jm00281a014>
30. **Heubach G., Sachse B., Buerstell H.** *Ger. Offen.* 826760, 1979.
31. **Tanaka G.** *Japan Kokai.* 973. 7495. 1974
32. **Griffin D. A., Mannion S. K.** *Eur. Pat. Appl.* EP 199,474, 1986.
33. **Hanna N. B., Dimitrijevič S. D., Larson S. B., Robsin R. K., Revankar G. R.** Synthesis and single-crystal X-ray diffraction studies of 1- β -D-ribofuranosyl-1,2,4-triazole-3-sulfonamide and certain related nucleosides // *J. Heterocycl. Chem.* 1988. Vol. 25. P. 1857–1868.
<http://dx.doi.org/10.1002/jhet.5570250649>
34. **Jenkins T. C., Stratford I. J., Stephens M. A.** 3-Nitro-1,2,4-triazoles as hypoxia-selective agents // *Anticancer Drug Des.* 1989. Vol. 4. P. 145–160.
35. **Husain M. I., Amir M.** Synthesis of some new substituted thiosemicarbazides and triazoles as possible anticonvulsants // *J. Indian Chem. Soc.* 1986, Vol. 63. P. 317–319.

36. **Chiu S.-H. L., Huskey S.-E. W.** Species differences in *N*-glucuronidation // *Drug Metabol. Dispos.* 1998. Vol. 26. P. 838–847.
37. **Elliott R., Sunley R. L., Griffin D. A.** PrepUK PatAppl GB 2 175 301, 1986.
38. **Lown J. W.** Anthracycline and anthracendione-based anticancer agents. Elsevier Science Publ. B. V. Amsterdam–Oxford–New York–Tokyo. 1988. 738 p.
39. **Delmulle L., Demeyer K.** Anthraquinones in plants: source, safety and applications in gastrointestinal health. Nottingham University Press. 2010. 157 p.
40. **Файн В. Я.** 9,10-Антрахиноны и их применение. Центр фотохимии РАН. Москва. 1999. 92 с.
41. **Poroikov V. V., Filimonov D. A.** How to acquire new biological activities in old compounds by computer prediction // *J. Comput. Aid. Molec. Des.* 2002. Vol. 11. P. 819–824.
<http://dx.doi.org/10.1023/A:1023836829456>
42. **Filimonov D. A., Poroikov V. V.** In: *Chemoinformatics Approaches to Virtual Screening*. Eds. Alexandre Varnek and Alexander Tropsha. Cambridge, RSC Publishing, 2008. P. 182–216.
<http://dx.doi.org/10.1039/9781847558879-00182>
43. URL [<http://www.way2drug.com/passonline>]
44. **Geronikaki A., Druzhilovsky D., Zakharov A., Poroikov V.** Computer-aided predictions for medicinal chemistry via Internet // *SAR and QSAR in Environ. Res.* 2008. Vol. 19. P. 27–38.
<http://dx.doi.org/10.1080/10629360701843649>
45. **Kryzhanovskii S. A., Salimov R. M., Lagunin A. A., Filimonov D. A., Gloriozova T. A., Poroikov V. V.** Nootropic action of some antihypertensive drugs: computer predicting and experimental testing // *Pharmaceut. Chem. J.* 2012. Vol. 45. 10. P. 605–611.
46. URL: [<http://integrity.thomson-pharma.com>]
47. **Stasevych M., Zvarych V., Musyanovych R., Novikov V., Vovk M.** Synthesis of *N*-benzoyl-*N'*-(9,10-dioxo-9,10-dihydroanthracene-1-yl)-thioureas and quantum-chemical analysis of the reaction passing // *Chemistry & Chemical Technology.* 2014. Vol. 3 (in press).
- V. Zvarych, M. Stasevych, O. Stanko, V. Novikov, M. Vovk, V. Poroikov, O. Solovyov

NAUJŲ N-BENZOIL-N'-(9,10-DIOKSO-9,10-DIHIDROANTRACEN-1-IL)TIOŠLAPALO PAGRINDU AZOLO DARINIŲ SINTEZĖ IR KOMPIUTERINIS JŲ AKTYVUMO VERTINIMAS

S a n t r a u k a

Nauji 9,10-antrachinono dariniai, turintys tiazolo arba 1,2,4-triazolo fragmentus, buvo susintetinti reaguojant *N*-benzoyl-*N'*-(9,10-diokso-9,10-dihydroantracene-1-il)tiošlapalo dariniams su bromacetonu, reakcijos terpėje esant trietilaminui arba hidrazino hidratui chloroforme. Panaudojant PASS kompiuterinę programą, buvo apskaičiuoti biologinio aktyvumo spektrai, siekiant atrinkti kiek įmanoma aktyviausius junginius biologinio aktyvumo tyrimams atlikti. Šie skaičiavimai parodė, kad susintetinti junginiai turėtų būti labai biologiškai aktyvūs.